

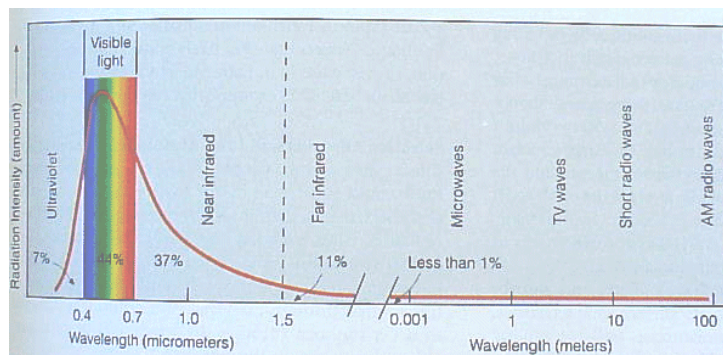
Radiation and the Planetary Energy Balance

- Electromagnetic Radiation
- Solar radiation warms the planet
- Conversion of solar energy at the surface
- Absorption and emission by the atmosphere
- The greenhouse effect
- Planetary energy balance

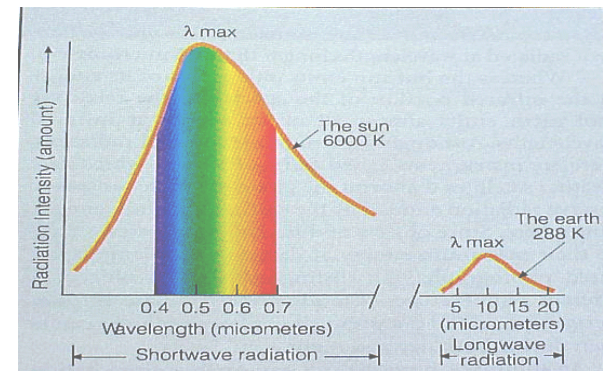
Electromagnetic Radiation

- Oscillating **electric and magnetic fields** propagate through space
- Virtually **all energy exchange between the Earth and the rest of the Universe** is by electromagnetic radiation
- Most of **what we perceive as temperature** is also due to our radiative environment
- May be described as **waves or as particles (photons)**
- **High energy photons = short waves;**
lower energy photons = longer waves

Electromagnetic Spectrum of the Sun



Spectrum of the sun compared with that of the earth



Ways to label radiation

- By its source
 - Solar radiation - originating from the sun
 - Terrestrial radiation - originating from the earth
- By its name
 - ultra violet, visible, near infrared, infrared, microwave, etc....
- By its wavelength
 - short wave radiation $\lambda < 3$ micrometers (μm)
 - long wave radiation $\lambda > 3$ micrometers

Absorption of Solar Radiation

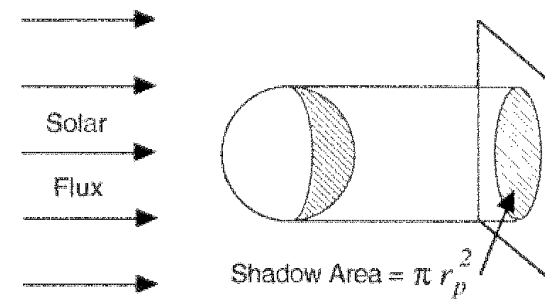


Fig. 2.2 Diagram showing the shadow area of a spherical planet.

Blackbodies and Graybodies

- A **blackbody** is a hypothetical object that **absorbs all of the radiation that strikes it**. It also emits radiation at a maximum rate for its given temperature.
 - Does not have to be black!
- A graybody absorbs radiation equally at all wavelengths, but at a **certain fraction (absorptivity, emissivity) of the blackbody rate**

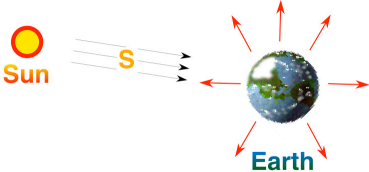
Total Blackbody Emission

- The **total rate of emission of radiant energy from a "blackbody"**:

$$E^* = \sigma T^4$$

- This is known as the **Stefan-Boltzmann Law**, and the constant σ is the Stefan-Boltzmann constant ($5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$).
- Stefan-Boltzmann says that **total emission depends really strongly on temperature!**
- This is strictly true only for a blackbody. For a **gray body**, $E = \epsilon E^*$, where ϵ is called the **emissivity**.
- In general, the **emissivity depends on wavelength** just as the absorptivity does, for the same reasons: $\epsilon_\lambda = E_\lambda / E^*_\lambda$

Planetary Energy Balance



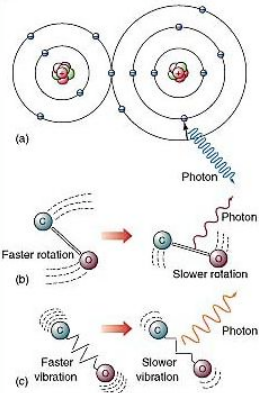
Energy In = Energy Out

$$S(1-\alpha)\pi R^2 = 4\pi R^2\sigma T^4$$

$$T \approx -18^\circ\text{C}$$

But the observed T_s is about 15°C

Atoms, Molecules, and Photons



- Atmospheric gases are made of molecules
- Molecules are groups of atoms that share electrons (bonds)
- Photons can interact with molecules
- Transitions between one state and another involve specific amounts of energy

Molecular Absorbers/Emitters

Molecule	Arrangement	Permanent Dipole Moment
N ₂		No
O ₂		No
CO		Yes
CO ₂		No
N ₂ O		Yes
H ₂ O		Yes
O ₃		Yes
CH ₄		No

Diatomic Structures

N₂, O₂, CO

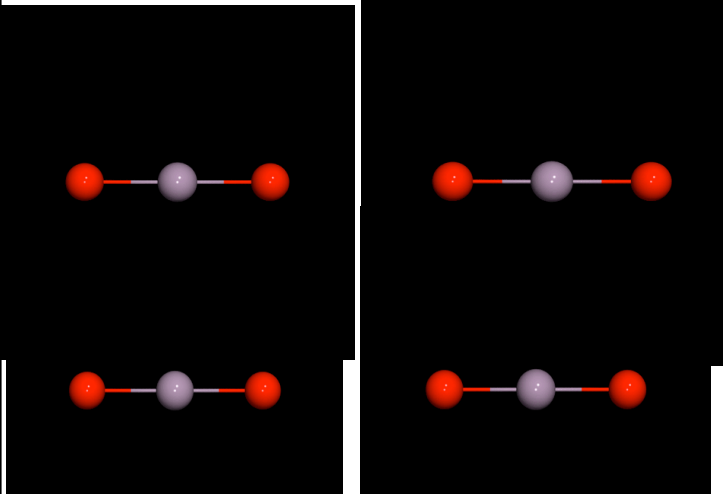
Triatomic Structures

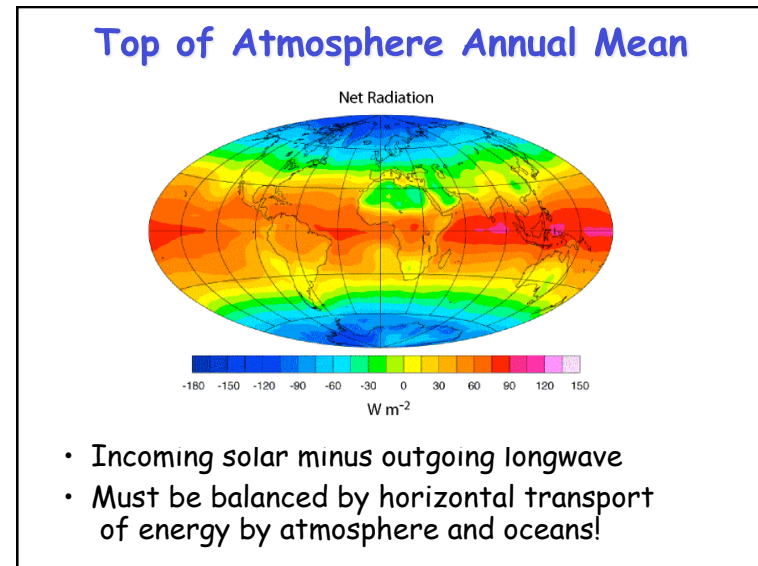
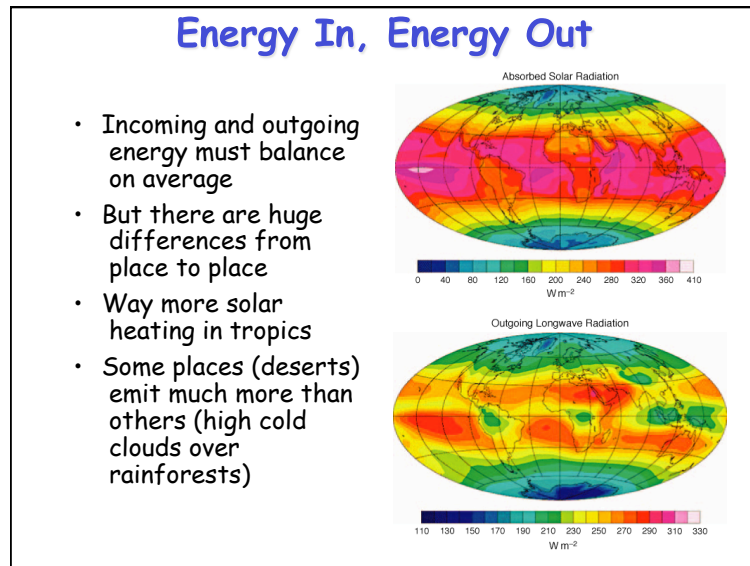
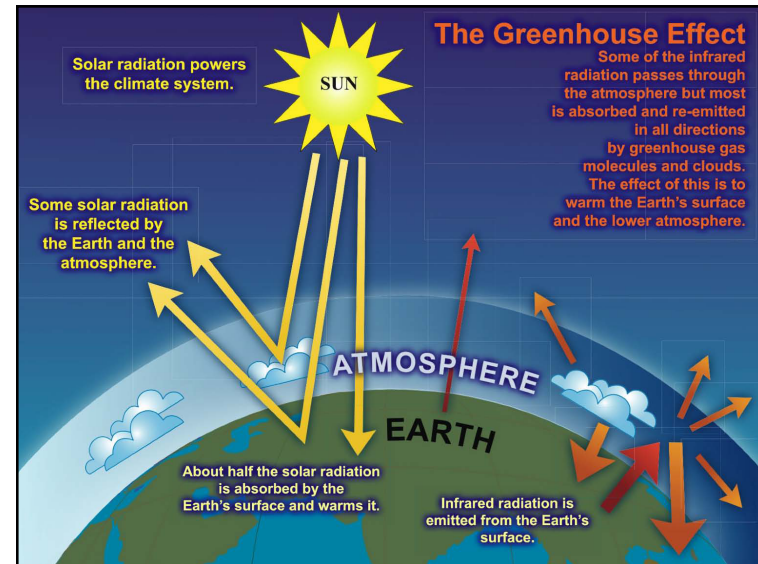
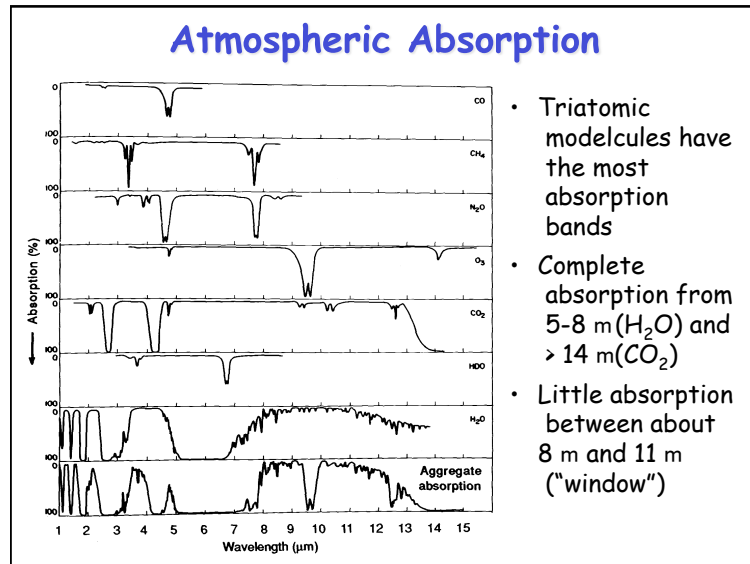
CO₂, N₂O

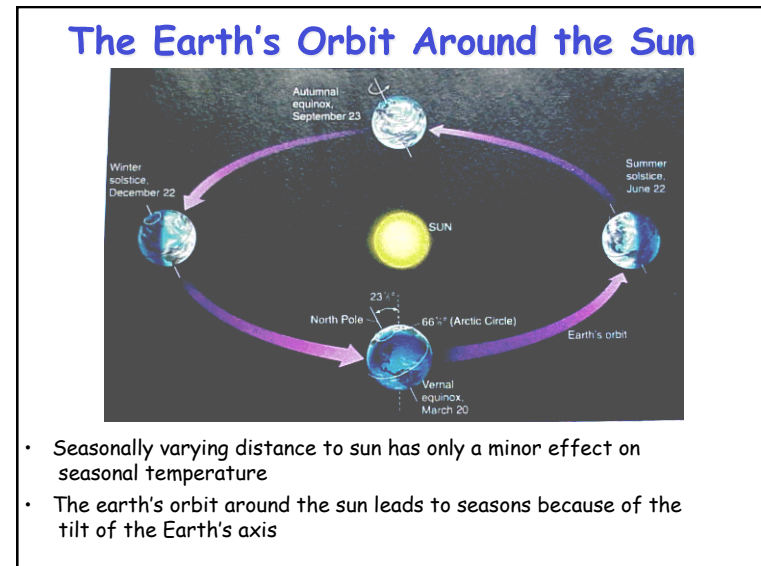
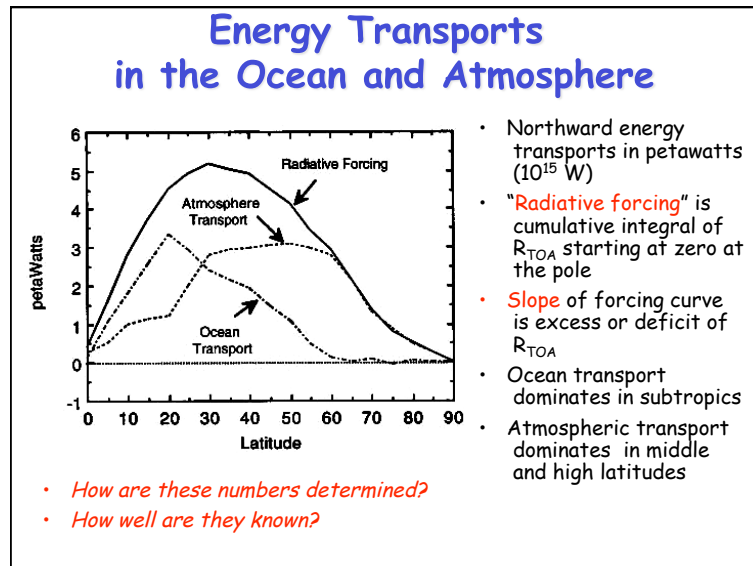
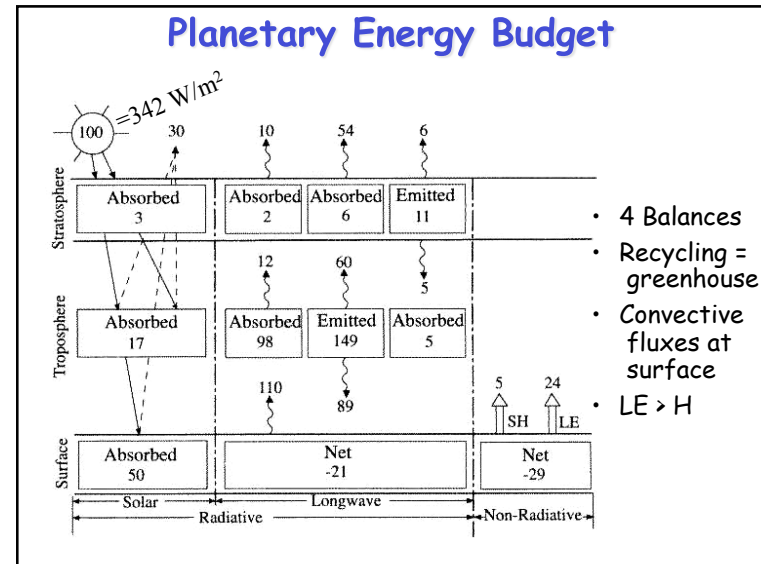
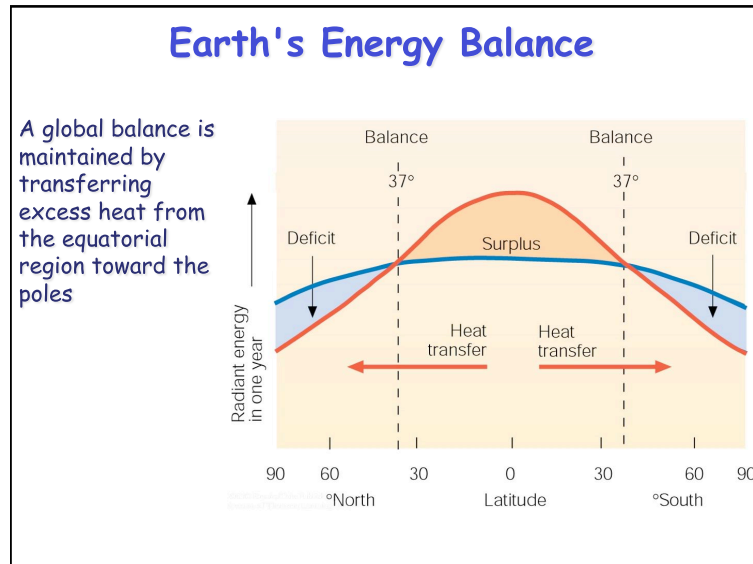
H₂O, O₃

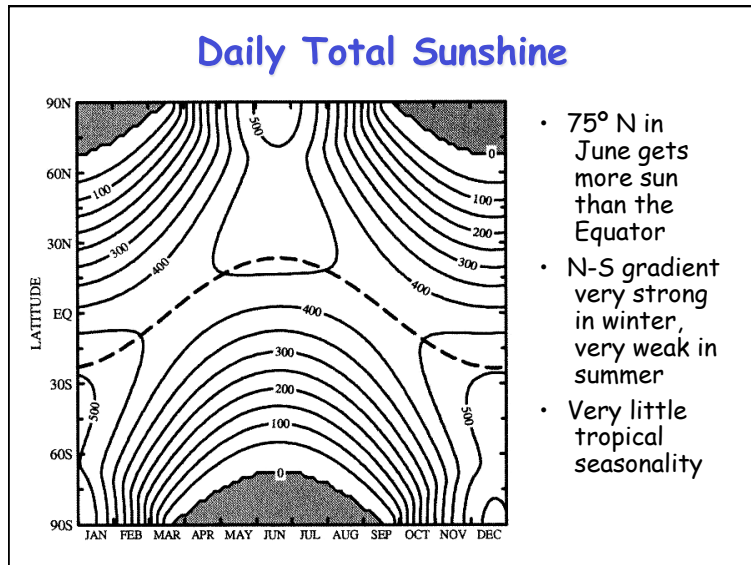
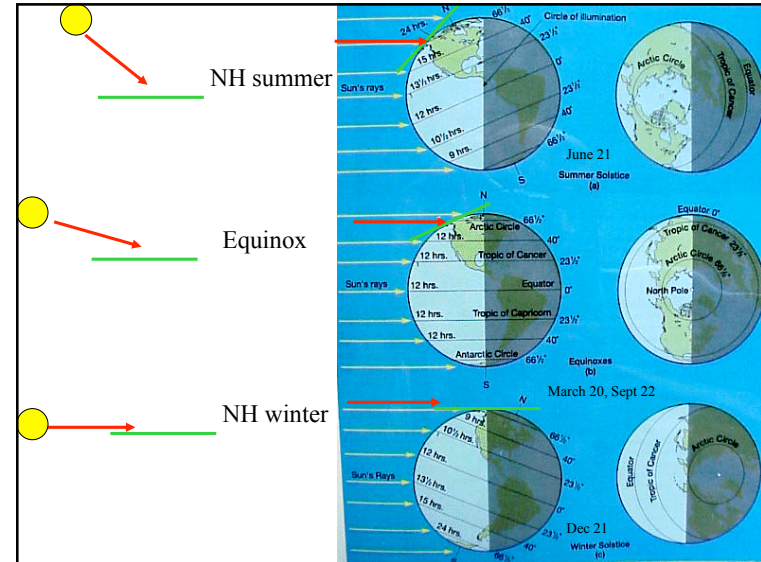
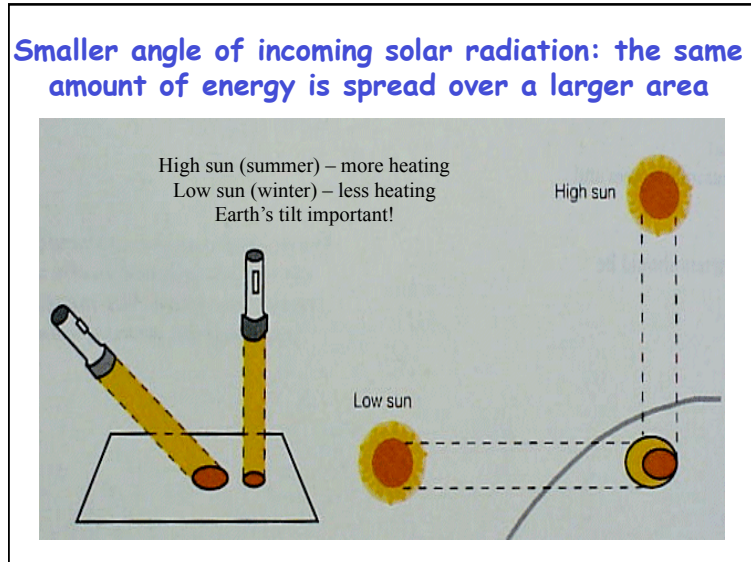
- **Molecules** of gas in the atmosphere interact with **photons** of electromagnetic radiation
- Different kinds of molecular transitions can absorb/emit very **different wavelengths** of radiation
- Some molecules are able to interact much more with photons than others
- Molecules with **more freedom to jiggle and bend** in different ways absorb more types of photons
- Water vapor (H₂O) and CO₂ are pretty good at this, and abundant enough to make a big difference!
- These are the **"greenhouse gases!"**

CO₂ Vibration Modes





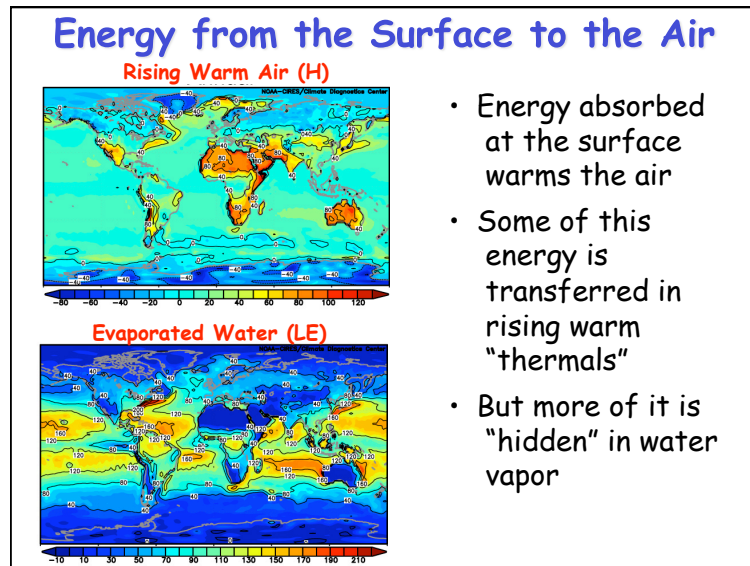
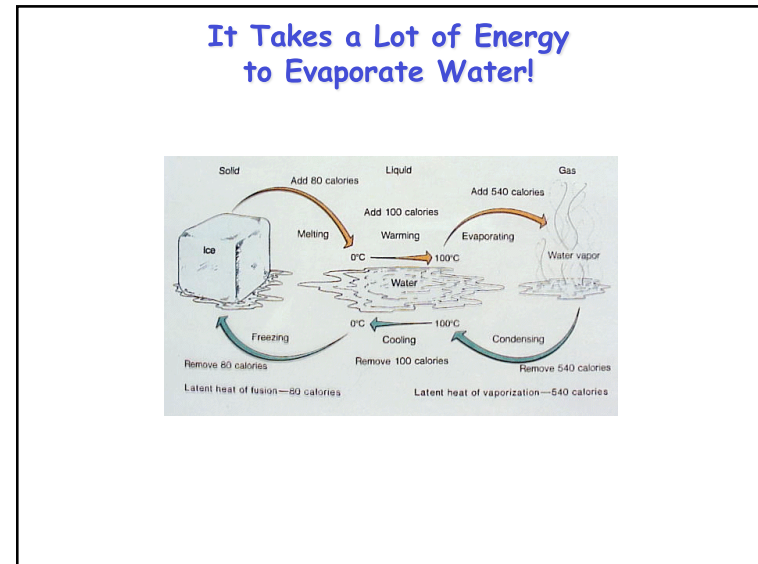
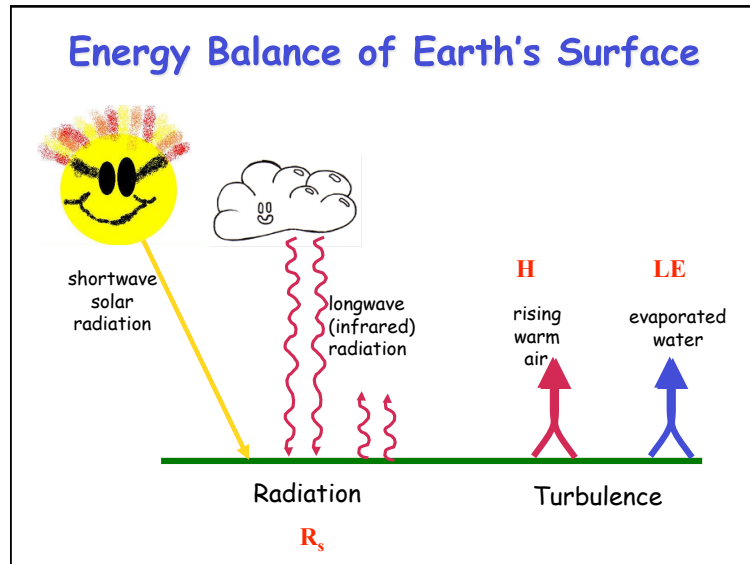




Surface Albedos (percent)

Surface type	Range	Typical value
Water		
Deep water: low wind, low altitude	5–10	7
Deep water: high wind, high altitude	10–20	12
Bare surfaces		
Moist dark soil, high humus	5–15	10
Moist gray soil	10–20	15
Dry soil, desert	20–35	30
Wet sand	20–30	25
Dry light sand	30–40	35
Asphalt pavement	5–10	7
Concrete pavement	15–35	20
Vegetation		
Short green vegetation	10–20	17
Dry vegetation	20–30	25
Coniferous forest	10–15	12
Deciduous forest	15–25	17
Snow and ice		
Forest with surface snowcover	20–35	25
Sea ice, no snowcover	25–40	30
Old, melting snow	35–65	50
Dry, cold snow	60–75	70
Fresh, dry snow	70–90	80

- Snow and ice brightest
- Deserts, dry soil, and dry grass are very bright
- Forests are dark
- Coniferous (cone-bearing) needleleaf trees are darkest



- ### Things to Remember
- All energy exchange with Earth is radiation
 - Outgoing radiation has longer waves (cooler)
 - Longwave radiation is absorbed and re-emitted by molecules in the air (H_2O & CO_2)
 - Recycling of energy between air and surface is the "greenhouse effect"
 - Changes of angle of incoming sunlight and length of day & night are responsible for seasons and for north-south differences in climate
 - Regional energy surpluses and deficits drive the atmosphere and ocean circulations (wind & currents)